

Romulus and Remus the twins of Cancer



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Introduction

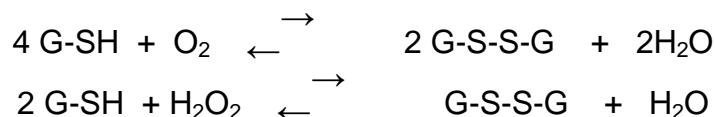
Twins are associated with Cancer for promotion of proliferation, drug and radiation resistance. The first twin *Romulus* is Lactic Acid and the second twin *Remus* is Glutathione. Together they are jointly necessary for supporting the increased rate of Cancer cell metabolism, progression, drug and radiation resistance.

Lactic Acid

Cancer malignancy is defined by an increase in the cellular metabolic rate with increased generation of Lactic acid which is transported outside the cells into the microenvironment. (11, 12, 14, 15, 18).

Glutathione

Cancer cells have elevated levels of Glutathione. (1,19, 20, 21, 22 ,23 ,25) One of the main functions of Glutathione in biological systems is to maintain the balance of the Reduction-Oxidation system (REDOX)



As Cancer cell metabolic rates increase there are proportionally more Reactive Oxygen Species (ROS) generated which needs to be balanced by higher levels of the antioxidant Glutathione in the Cancer cells. Excessive ROS levels can cause cellular damage and cell death but low levels of ROS are important for biological system operation and viability. Maintaining this balance is critical for cell survival and proliferation. (1, 3, 23, 25, 29)

Cancer Cell Glutathione Elevation and Drug-Radiation Resistance

Cancer cells as they become more malignant have increased Glutathione levels which contribute to increased drug and radiation resistance. (1, 16, 19, 20, 21, 22)

Increased resistance to drugs such as Cisplatin and Arsenic Oxide has been shown to be directly proportional to the Cancer cell Gutathione level. (1, 16, 19, 20, 21, 22).

The Glutathione can have multiple modes of action. Glutathione like most compounds bearing a Thiol group can act as a simple antioxidant, metal chelator or bind to any active enzyme or chemical with a Thiol group. (1, 4, 19, 20, 21, 22).

Glutathione also forms numerous conjugates in biological systems by other binding mechanisms (29) which can lead to Multi Drug Resistance.

Glutathione Decline in the Cancer Microenvironment.

Increasing levels of Glutathione in the Cancer cells accompanies a lowering of Glutathione in the microenvironment. Cancer patients typically have declining Blood Plasma Glutathione levels. (23, 24, 25)

Proton Pump Inhibitor Effects on Lactic Acid levels, Glutathione Levels and Cancer Progression

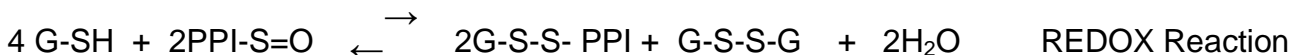
Inhibition of the Cancer cell Proton Pumps will lower Lactic acid production in a dose dependent manner. This lowering of the Lactic acid production will eventually lower Glutathione levels in the cancer cell. The Cancer cell metabolic rate will drop resulting in a decline in the Cancer growth rate.

Proton Pump Inhibitor Thiol Interactions and Di-Sulphide Formation

The PPIs are Pro Drugs and oxidants which can be reduced by other biological Thiol based compounds to form Di-Sulphide compounds. (4, 26, 27). These Di-Sulphide compounds can be involved with Thiol – Di-Sulphide reversible exchanges which are a regular feature in biological systems. The Thiol group of the Cancer cell Proton Pump will be only one that PPIs can react with in competition with other Thiol based compounds. The PPI Thiol group can also bind to drugs such as Cisplatin which mitigates against co-administration.

The Link between Proton Pump Inhibitor Transport, Stabilization and Plasma Glutathione Levels

PPIs are acid labile. In the presence of acid and absence of Thiol compounds PPIs break down to inactive compounds. PPIs transported through Cancer acidic microenvironments before reaching the Cancer cells can suffer acidic degradation if there is insufficient Glutathione present to form a stable Glutathione--PPI Di-Sulphide compound.



Glutathione and other low molecular weight Thiol compounds in Plasma can stabilize and increase the transport efficiency of the PPI through the acidic Cancer cell microenvironment. (4, 6, 10, 13). Once transported through the acidic microenvironment the Glutathione-PPI Di-Sulphide is free to undergo Di-Sulphide exchange with the Cancer cell proton pumps (10). This Di-Sulphide exchange is an equilibria reversible reaction.

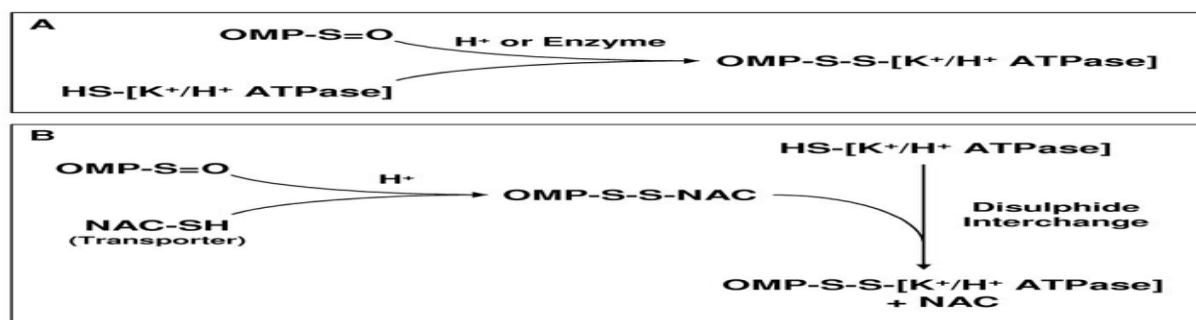
Based on Cancer cell culture trials stabilizing the PPI as a Di-Sulphide shows better PPI transport and response with a range of Cancer cell Types. (10,13)



Metabolic Approaches to Treatment of Melanoma

Clin Cancer Res November 1, 2009 15; 6490

Peter Hersey, Ralph Neal Watts, Xu Dong Zhang and John Hackett



OMP-S=O is Omeprazole
NAC-SH is N-Acetyl Cysteine
HS-[K⁺/H⁺ ATPase] is a Proton Pump

FIGURE 2

A. Omeprazole binding to sulfhydryl groups on the proton pumps.

B. Role of N-Acetyl Cysteine in acting as a transporter molecule for Omeprazole to proton pumps in cancer cells.

The reverse reaction of the PPIs being displaced from the Gastric Proton Pump by Glutathione has been noted and measured. (28). Lansoprazole showed a Half Life of less than 15 hours, and both Omeprazole and Rabeprazole showed less than 30 Hours, for Pantoprazole the Half Life was approximately 46 hours . The longer Half Life of Pantoprazole is attributed to the inability of Glutathione to reach the less exposed Thiol sites targeted by Pantoprazole.

If Cancer cell Proton Pumps responded in a similar way one might expect Pantoprazole to be a better choice for the treatment of cancers but not necessarily so if a particular type of Cancer Proton Pump has a different design to that of the Gastric Proton Pump.

Glutathione Levels in Laboratory Models compared with Clinical Patients.

Most laboratory Cancer cell culture and animal models invariably have higher Glutathione levels in the Cancer microenvironment compared with Cancer patients. Healthy rats as an example have about fifteen times the plasma Glutathione level than humans . (23, 24, 25). PPI Cancer Clinical trials involving subjects with depressed plasma Glutathione levels will not translate to the better results obtained in laboratory studies with higher plasma Glutathione levels. (7, 2)

Elderly subjects both human and animal have a lower Plasma Glutathione level (24). They are more prone to cancers and refractory to treatment. Less efficient PPI transport in these patients will compromise treatment. Variation in the Plasma Glutathione level in the general population and Cancer patients is also an important factor in the effectiveness of treatments using PPIs for individual patients.

Multifunction Role of Proton Pump Inhibitors

The PPIs do not only inhibit Proton pumps but can inhibit other metabolic pathways in a dose dependent manner (6).

- (a) Proton Pumps $H^+K^+ATPase$, $H+Na+ATPase$, $H+Ca^{2+}+ATPase$.
- (b) Carbonic Anhydrase
- (c) Pyruvate Decarboxylase
- (d) Transketolase
- (e) Histone Deacetylase
- (f) Superoxide Dismutase SOD
- (g) Zinc Metallo Hydrolase Family
- (h) Zinc and Copper based enzymes

. At higher dose rates than those required for Proton Pump inhibition other metabolic pathways can be inhibited to cause cancer cell death.

Increasing PPI Transport in Patient Clinical Trials

It has proven difficult to increase Cancer Patient Plasma Glutathione level which would improve PPI transport.

An alternate approach is to utilize the N-Acetyl Cysteine–PPI Di-Sulphide which has performed better in Cancer cell culture studies by a factor of three. (6,13). It has a lower molecular weight than the equivalent Glutathione–PPI Di-Sulphide with improved drug transport and is capable of passing the blood-brain barrier. (4, 10, 13)

Conclusion.

Proton Pump inhibitors have an important role to play in the treatment of Cancers. Clinical trials to date have been hampered by inefficient Proton Pump Inhibitor transport and insufficient recognition of the joint Lactic acid and Glutathione roles in Cancer progression and treatment.

Foot Note;

Apologies are extended to those Researchers whose references could not be included in this brief Review.

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